## Stereospecific Synthesis of (E)-Vinyl Sulphides via 1-lodoalk-1-enyldialkylboranes

Masayuki Hoshi, Yuzuru Masuda, and Akira Arase\* Department of Industrial Chemistry, Kitami Institute of Technology, Kitami 090, Japan

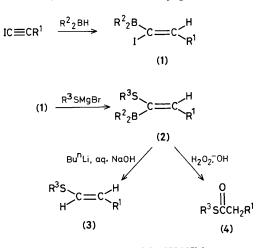
Isomerically pure (*E*)-vinyl sulphides were obtained by reaction of 1-iodoalk-1-enylbis(1,2-dimethylpropyl)boranes successively with alkylthiomagnesium bromides, n-butyl-lithium, and aqueous sodium hydroxide.

Vinyl sulphides are potentially useful intermediates in organic synthesis,<sup>1</sup> and their stereospecific synthesis has been attempted. We report a new, highly stereospecific synthesis of (E)-vinyl sulphides (3) using 1-iodoalk-1-enylbis(1,2-dimethylpropyl)boranes (1).

We previously found that compounds (1) were transformed into S-alkyl alkanethioates (4) by reaction with alkylthiomagnesium bromides followed by alkaline hydrogen peroxide.<sup>2</sup> The 1-alkylthioalk-1-enylbis(1,2-dimethylpropyl)boranes (2) which we assume are formed as intermediates would afford vinyl sulphides by protonolysis.

Preliminary g.l.c. examination of the *in situ* protonolysis of the mixtures obtained by the reaction of (1) with alkylthiomagnesium bromides,<sup>†</sup> using carboxylic acids, normal reagents for the protonolysis of alkenylboranes,<sup>3</sup> failed to

<sup>†</sup> Alkylthiomagnesium bromides were prepared by the slow addition of alkanethiols to ethylmagnesium bromide in tetrahydrofuran.



Scheme 1.  $R^2 = Me_2CHCHMe$ 

show any detectable products. However, treatment of the same mixtures with n-butyl-lithium and then aqueous sodium hydroxide, a method reported by Negishi *et al.* for conversion of the carbon-boron bond of alk-1-enylboranes into a carbon-hydrogen bond with retention of configuration,<sup>4</sup> afforded (3) in about 50% yields based on the 1-iodoalk-1-yne precursors of (1). Addition of hexamethylphosphoric triamide (HMPT) as a co-solvent increased the yields markedly.

As shown in Table 1, the reaction could be applied to various organothiomagnesium bromides including alkylthio-, phenylthio-, and benzylthio-compounds. Isomerically pure compounds (3) were isolated following work-up by column chromatography using basic aluminium oxide coated with sodium hydroxide.<sup>‡</sup> The configuration was assigned to all compounds (3) by the coupling constant (J 14.7—15.1 Hz) for the vinyl protons in the <sup>1</sup>H n.m.r. spectra. Signals resulting from the Z isomers were not found in any cases, indicating that the reactions were stereospecific.

Previously, Vermeer *et al.* reported a highly stereoselective synthesis of (*E*)-vinyl sulphides employing alk-1-ynyl sulphides  $RC\equiv CSR'$  as the starting materials.<sup>5</sup>

 $\ddagger$  Aluminium oxide was treated with 10% aqueous NaOH and dried at 150 °C.

Table 1. Yields of (E)-vinyl sulphides (3).<sup>a</sup>

R <sup>3</sup>	Yield of (3) <sup>b</sup>
Bu <sup>n</sup>	82 (45)
But	87
Ph	67
$PhCH_2$	62
Bu <sup>n</sup>	70
	Bu <sup>n</sup> Bu <sup>t</sup> Ph PhCH <sub>2</sub>

<sup>a</sup> The reactions of R<sup>1</sup>CH=CIBR<sup>2</sup><sub>2</sub> (1), prepared by hydroboration of 1-iodoalk-1-ynes (10 mmol) with bis(1,2-dimethylpropyl)borane (10 mmol) in tetrahydrofuran (THF), with R<sup>3</sup>SMgBr (10 mmol) in THF, were carried out at -50 °C for 1 h followed by warming to room temperature, addition of HMPT (5 ml), and then treatment with Bu<sup>n</sup>Li (10 mmol) in hexane and aqueous 6 M-NaOH (10 ml). <sup>b</sup> Isolated yields based on starting 1-iodoalk-1-yne; yield in parentheses in the absence of HMPT.

The configurations of compounds (2) could not be defined directly. However, our results strongly suggest that the alkylthio groups were introduced at the  $\alpha$ -alkenyl carbon atoms of compounds (1) with inversion of configuration giving the (E)-alkenylboranes (2).

The advantages of the present synthesis are that all reagents, including 1-iodoalk-1-ynes, are readily available and relatively cheap, and all reactions may be carried out without isolation of the intermediates (2), giving isomerically pure products (3).

Received, 8th May 1985; Com. 623

## References

- 1 For example: G. C. Barrett, in 'Comprehensive Organic Chemistry,' eds. D. Barton and W. D. Ollis, Pergamon, Oxford, 1979, vol. 3, p. 78.
- 2 M. Hoshi, Y. Masuda, and A. Arase, J. Chem. Soc., Chem. Commun., 1985, 714.
- 3 H. C. Brown and G. Zweifel, J. Am. Chem. Soc., 1961, 83, 3834.
- 4 E. Negishi and K. W. Chiu, J. Org. Chem., 1976, 41, 3484.
- 5 P. Vermeer, J. Meijer, C. Eylander, and L. Brandsma, *Recl. Trav. Chim. Pays-Bas*, 1976, **95**, 25.